

# Structure of the exceptionally large nonrepetitive carbohydrate backbone of the lipopolysaccharide of *Pectinatus frisingensis* strain VTT E-82164

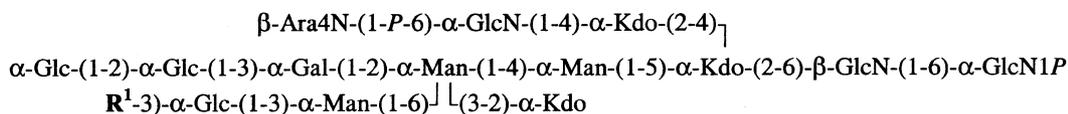
Evgeny Vinogradov<sup>1</sup>, Bent O. Petersen<sup>2</sup>, Irina Sadovskaya<sup>3</sup>, Said Jabbouri<sup>3</sup>, Jens Ø. Duus<sup>2</sup> and Ilkka M. Helander<sup>4</sup>

<sup>1</sup>Institute for Biological Sciences, National Research Council, Ottawa, ON, Canada; <sup>2</sup>Department of Chemistry, Carlsberg Laboratory, Copenhagen, Denmark; <sup>3</sup>Laboratoire de Recherche sur les Biomatériaux et Biotechnologies, Université de Littoral-Côte d'Opale, Bassin Napoléon BP 120, Boulogne-sur-mer, France; <sup>4</sup>Department of Applied Chemistry and Microbiology, Division of Microbiology, University of Helsinki, Finland

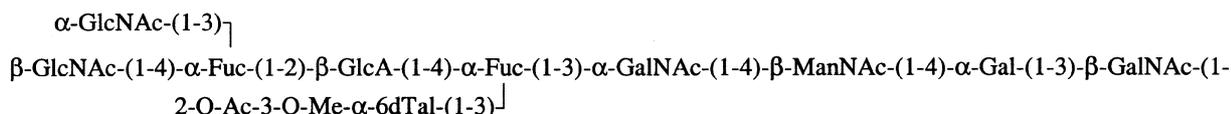
The structures of the oligosaccharides obtained after acetic acid hydrolysis and alkaline deacylation of the rough-type lipopolysaccharide (LPS) from *Pectinatus frisingensis* strain VTT E-82164 were analysed using NMR spectroscopy, MS and chemical methods. The LPS contains two major structural variants, differing by a decasaccharide fragment, and some minor variants lacking the terminal glucose residue. The largest structure of the carbohydrate backbone of the LPS that could be deduced from experimental results consists of 25 monosaccharides (including the previously found Ara4NP residue in lipid A) arranged in a well-defined non-repetitive structure:

We presume that the shorter variant with  $R^1 = H$  represents the core-lipid A part of the LPS, and the additional fragment is present instead of the O-specific polysaccharide. Structures of this type have not been previously described. Analysis of the deacylation products obtained from the LPS of the smooth strain, VTT E-79100<sup>T</sup>, showed that it contains a very similar core but with one different glycosidic linkage.

**Keywords:** core; lipid A; lipopolysaccharide; *Pectinatus frisingensis*.



where  $R^1$  is H or



Strictly anaerobic Gram-negative rod-shaped bacteria causing turbidity and off flavours in bottled beer were initially isolated in 1978 and described as *Pectinatus cerevisiophilus*

[1]. Another species, *Pectinatus frisingensis*, which differed from *P. cerevisiophilus* in a number of biochemical characteristics was later described [2]. To date, the VTT culture collection (Espoo, Finland) has 32 *Pectinatus* isolates from spoiled beer originating from Belgium, Finland, Germany, the Netherlands and the USA; 24 have been identified as *P. frisingensis* and eight as *P. cerevisiophilus* by conventional tests, ribotyping and partial 16S rDNA sequence analysis [3].

The lipopolysaccharides (LPS) of type strains of *P. cerevisiophilus* and *P. frisingensis* possess a number of remarkable properties, including the predominance of odd-numbered fatty acids in lipid A [4] and the presence of furanosidic 6-deoxysugars in the O-specific chains [5]. The lipid A was shown to be quantitatively substituted at the 4'-phosphate and partially at the glycosidic phosphate by 4-amino-4-deoxy- $\beta$ -L-arabinose [6]. There are no structural data on

Correspondence to E. Vinogradov, Institute for Biological Sciences, National Research Council, 100 Sussex Dr, K1A 0R6 Ottawa ON, Canada.

Fax: +1 613 952 90 92, Tel.: +1 613 990 03 97,

E-mail: evgenii.vinogradov@nrc-cnrc.gc.ca

Abbreviations: LPS, lipopolysaccharide; Kdo, 3-deoxy-D-manno-oct-2-ulosonic acid; Ara4N, 4-amino-4-deoxy-L-arabinose;

HPAEC, high-performance anion-exchange chromatography;

ESI MS, electrospray ionization mass spectrometry.

(Received 5 April 2003, revised 14 May 2003,

accepted 22 May 2003)

*Pectinatus* core structures, except a report that LPS of both *P. frisingensis* and *P. cerevisiophilus* contain a disaccharide structure, a phosphorylated GlcN linked to O4 of a Kdo residue, tentatively assigned to the core region [7].

Screening of *Pectinatus* strains other than type strains has revealed that the LPS from certain strains exhibit only two distinct bands on PAGE, with no polymeric O chains (I. M. Helander, unpublished data). This indicates the presence of two structurally distinct LPS molecules. We describe here the chemical structure of the LPS carbohydrate backbone of one such isolate, *P. frisingensis* VTT-E-82164, which has 99.8% similarity of partial 16S rDNA to the *P. frisingensis* type strain.

## Materials and methods

### Bacterial strains and growth conditions

*P. frisingensis* VTT E-82164 and VTT E-79100<sup>T</sup> and *P. cerevisiophilus* E-79103<sup>T</sup> were obtained from VTT Biotechnology (Espoo, Finland) [3]. Cells were grown anaerobically at 32 °C without shaking in Man Rogosa Sharpe broth (Difco), pH 6.5, in the presence of a reducing agent (Na<sub>2</sub>S, 12.5 mM) and resazurin (1 mg·mL<sup>-1</sup>), and collected at the stationary growth phase.

### LPS isolation

Bacterial cells were washed with ethanol, acetone, and light petroleum, and LPS was extracted from the dried cells with phenol/chloroform/petroleum ether (60–95 °C) (5 : 5 : 8, v/v) with acetone precipitation [4,8].

### NMR spectroscopy and general methods

NMR spectra were recorded at 25 °C in D<sub>2</sub>O on a Varian Unity Inova 800 instrument at 799.96 MHz for proton and 201.12 MHz for carbon, using acetone as reference for proton (2.225 p.p.m.) and 1,4-dioxane for carbon (67.4 p.p.m.). Varian standard programs tndqcosy, tnoesy (mixing time of 100 ms), tntocsy (spinlock time 80 ms), gHSQC, gHSQCTOCSY (spinlock time 80 ms), gHSQCNOESY (mixing time 200 ms) and gHMBC were used with digital resolution in F2 dimension < 2 Hz·pt<sup>-1</sup>. Spectra were assigned using the computer program PRONTO [9].

### Analytical methods

PAGE was performed with deoxycholate as the detergent. The separation gel contained 18% acrylamide, 0.5% (w/v) deoxycholate, and 375 mM Tris/HCl, pH 8.8, and stacking gel contained 4% acrylamide and 127 mM Tris/HCl, pH 6.8. LPS samples were prepared at a concentration of 0.1% (w/v) in sample buffer [127 mM Tris/HCl, pH 6.8, 10% (v/v) glycerol, 0.025% (w/v) bromphenol blue dye]. The electrode buffer was composed of deoxycholate (2.5 g·L<sup>-1</sup>), glycine (21.7 g·L<sup>-1</sup>), and Tris (4.5 g·L<sup>-1</sup>). Electrophoresis was performed at a constant current of 15 mA per gel with cooling. Immediately after the electrophoresis run, the gel was soaked in the fixing solution containing ethanol (40%, w/w) and acetic acid (5%, w/w). The solution

was changed after 30 min, and fixation continued overnight. LPS bands were visualized by silver staining as described by Tsai & Frasch [10].

Hydrolysis was performed with 4 M trifluoroacetic acid (110 °C, 3 h). Monosaccharides were conventionally converted into the alditol acetates and analysed by GC on a Agilent 6850 chromatograph equipped with a DB-17 fused-silica column (30 m × 0.25 mm) using a temperature gradient of 180 °C (2 min) → 240 °C at 2 °C·min<sup>-1</sup>. For the determination of the absolute configuration of 3-*O*-methyl-6-deoxytalose, GC was performed in isothermal conditions at 150 °C. GC-MS was performed on a Varian Saturn 2000 system with ion-trap mass spectral detector using the same column. Electrospray ionization (ESI) MS was carried out as described previously [11].

Gel chromatography was carried out on columns (2.5 × 95 cm) of Sephadex G-50 in pyridinium/acetate buffer, pH 4.5 (4 mL pyridine and 10 mL acetic acid in 1 L water) and BioGel P4 (1 × 90 cm) in water. The eluate was monitored with a refractive index detector.

Methylation analysis was performed by the Ciucanu-Kerek procedure [12]. Methylated products were hydrolysed and monosaccharides converted into 1d-alditol acetates by conventional methods and analysed by GC-MS.

High-performance anion-exchange chromatography (HPAEC) was performed on a CarboPac PA1 column (9 × 250 mm) with pulsed amperometric detection, equilibrated in 0.1 M NaOH, using a linear gradient of 1 M sodium acetate in 0.1 M NaOH from 5% to 80% of acetate in 60 min at 3 mL·min<sup>-1</sup>. Fractions of volume 3 mL were collected and analysed using the Dionex system with an analytical CarboPac PA1 column (4.6 × 250 mm) at 1 mL·min<sup>-1</sup>. Separated oligosaccharides were desalted on a Sephadex G-50 column.

### De-O,N-acylation of LPS and preparation of backbone oligosaccharides [13]

LPS (120 mg) was dissolved in 4 M KOH (4 mL), and the solution was heated at 120 °C for 16 h, cooled, neutralized with 2 M HCl. The precipitate was removed by centrifugation, and the supernatant desalted by gel chromatography on Sephadex G-50. Two oligosaccharide fractions with *K*<sub>av</sub> 0.60 and 0.47 were obtained and further separated by HPAEC on a semipreparative CarboPac PA1 column to give oligosaccharides **1a**, **1b** and a mixture of **2** and **3**.

### Deamination of the de-O,N-acylated LPS and preparation of oligosaccharides **4** and **5**

The mixture of oligosaccharides obtained after alkaline deacylation of the LPS (200 mg) was treated with 300 mg NaNO<sub>2</sub> in 10% acetic acid (10 mL, 25 °C, 24 h), desalted on a Sephadex G-50 column, reduced with NaBH<sub>4</sub>, desalted, and oligosaccharides **4** and **5** isolated by HPAEC.

### Acetic acid hydrolysis of LPS

LPS (100 mg) was treated with 2% acetic acid (5 mL, 100 °C, 3 h). The precipitate was removed by centrifugation, and the soluble products were separated on a Sephadex

G-50 column to give three oligosaccharide fractions. These were NaBH<sub>4</sub>-reduced, desalted, and separated by HPAEC to give oligosaccharides 6–8.

### Isolation of 3-*O*-methyl-6-deoxy-D-talose (11)

LPS (300 mg) was hydrolysed with 3 M trifluoroacetic acid (100 °C, 1.5 h), and the cooled dark solution was treated with activated carbon, filtered, and evaporated to dryness. An aqueous solution of the residue was passed through a column (0.8 × 15 cm) of Dowex 50W8 (× 200; H<sup>+</sup>), then through a column of Dowex 2 (AcO<sup>-</sup>). The monosaccharides were separated by paper chromatography on Whatman 3MM paper in pyridine/butanol/water (6 : 4 : 3, v/v/v). Sugars were detected on a small strip with AgNO<sub>3</sub>/NaOH reagent, and eluted with water. The portions of the fractions mainly containing Man, Glc, Gal, Fuc, and pure 3-*O*-methyl-6-deoxy-D-talose were treated with (*S*)-2-butanol/acetyl chloride (10 : 1, v/v; 2 h; 85 °C), dried under a stream of air, acetylated, and analysed by GC. 3-*O*-Methyl-6-deoxy-D-talose (3 mg) was obtained in pure form (moves close to front on paper); [α]<sub>D</sub><sup>20</sup> (c 0.3, water), lit. for L-isomer (trivial name acovenose) –14.2° (c 1.2, water) [14].

Amino sugars were eluted from Dowex 50 with 0.5 M HCl, N-acetylated (5 mL saturated NaHCO<sub>3</sub>, 0.5 mL acetic anhydride; 20°, 1 h with stirring), converted into (*S*)-2-butyl glycoside acetates as described above, and analysed by GC.

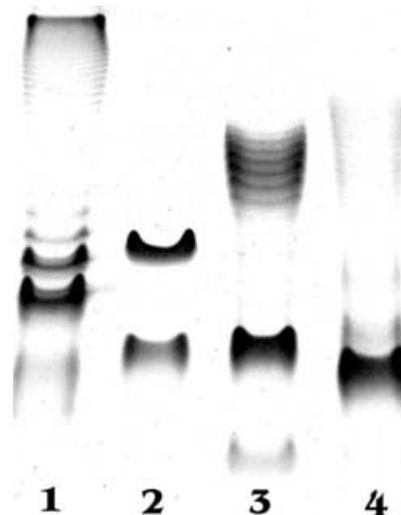
### Synthesis of methyl 3-*O*-methyl-6-deoxy-α-D-talopyranoside (9) and methyl 3-*O*-methyl-6-deoxy-β-D-talopyranoside (10)

3-*O*-Methyl-D-glucose (a gift from M. Perry, NRC Canada) (500 mg) was converted into an approximately 4 : 1 mixture of α-methyl and β-methyl glycosides by methanolysis (1 M HCl/MeOH; 85 °C; 24 h), brominated at C6 using CBr<sub>4</sub>/imidazole/triphenyl phosphine (1 : 1 : 2.5, v/v; 16 h; 25 °C; product isolated by column chromatography on SiO<sub>2</sub> in 5% MeOH in CHCl<sub>3</sub>), and debrominated by hydrogenolysis over Pd/C in MeOH to yield methyl 3-*O*-methyl-6-deoxy-α,β-glucopyranosides. These were converted into methyl 3-*O*-methyl-2,4-di-*O*-trifluorosulfonyl-6-deoxy-α,β-D-glucopyranoside [(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>O/Py; –20 °C to +25 °C] and treated with excess Et<sub>4</sub>NOAc in dimethylformamide (100 °C; 3 h). The reaction mixture was diluted 10 times with water, passed through Dowex 50 (H<sup>+</sup>) to remove Et<sub>4</sub>N<sup>+</sup>, evaporated to dryness, and compounds 9 and 10 were isolated by C<sub>18</sub> RP-HPLC in water (45 and 8 mg, respectively).

## Results

The LPS from *P. frisingensis* VTT E-82164 did not exhibit the typical ladder-like pattern of smooth LPS on deoxycholate-PAGE, but showed two main strongly stained rapidly migrating bands (Fig. 1).

Monosaccharide analysis of the whole LPS indicated the presence of fucose, 3-*O*-methyl-6-deoxyhexose, glucose, galactose, mannose, glucosamine, galactosamine, and mannosamine in the proportions 1 : 0.6 : 1.5 : 1.2 : 1.4 : 1.5 : 0.6 : 0.4.



**Fig. 1.** Deoxycholate-PAGE profiles of the LPS. Lane 1, *Salmonella enteritidis*; lane 2, *P. frisingensis* E-82164; lane 3, *P. frisingensis* type strain E-79100T; lane 4, *P. cerevisiphilus* type strain E-79103T.

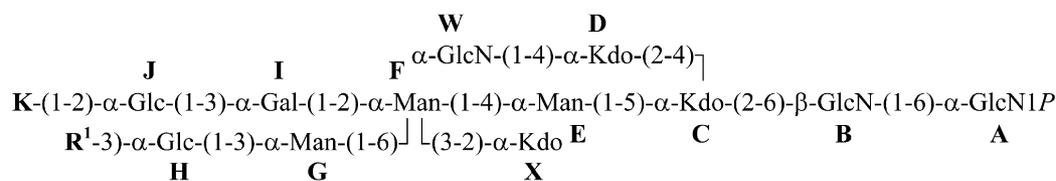
The LPS was O,N-deacylated by strong alkaline treatment. Gel chromatographic separation of the products on Sephadex G-50 gave two main peaks, which were further separated by HPAEC to give oligosaccharides 1a, 1b, and a mixture of 2 and 3 (Scheme 1).

In another experiment, the oligosaccharides obtained after deacylation and Sephadex G-50 separation were deaminated with nitrous acid and reduced with NaBH<sub>4</sub>. This led to removal of all amino sugar residues except B and O, which were transformed into 2,5-anhydromannitol and 2,5-anhydrotalitol, respectively. The products were separated by HPAEC and the oligosaccharides 4 and 5 were isolated.

Mild hydrolysis of the LPS with acetic acid and subsequent separation of the products by gel chromatography gave three oligosaccharide fractions. These were reduced with NaBH<sub>4</sub> and purified by HPAEC to give oligosaccharides 6, 7a, and 8. The longer oligosaccharide 7b was also analysed by NMR without reduction and HPAEC, which allowed detection of O-acetylation. Separation by HPAEC led to O-deacetylation of the reduced oligosaccharide 7b because of the alkaline chromatography conditions.

In 1D and 2D NMR spectra of compound 1a (Fig. 2), spin systems of 13 monosaccharides were identified. These were three α-Glc residues (H, J, K), three α-Man residues (E, F, G), one α-Gal residue (I), two α-GlcN (A, W) residues, one β-GlcN (B) residue, and three Kdo residues (C, D, X). The spectra were completely assigned (Table 1), and the sequence of the hexoses was determined from NOE and HMBC data, in which all respective strong transglycosidic correlations were observed. Assignments were made using methodology outlined in [15]. Oligosaccharides 1a and 1b are the fragments of the larger structure 2, thus the NMR data for 1a,b are very close to those for the respective residues in 2 and are not presented.

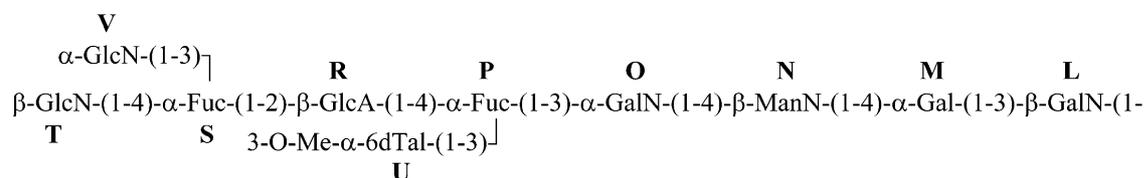
The position and anomeric configuration of Kdo residues was not as easy to assign. The <sup>1</sup>H and <sup>13</sup>C chemical shifts of Kdo residues C and X agreed well with



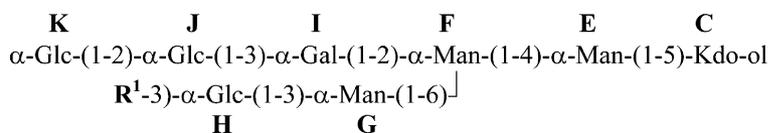
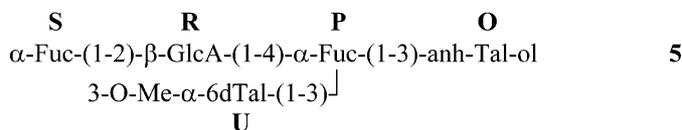
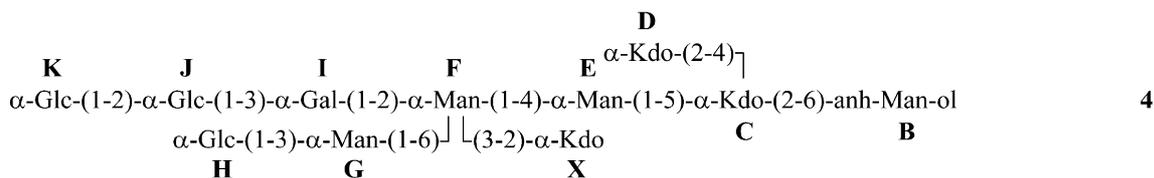
1a:  $\mathbf{K} = \alpha\text{-Glc}$ ,  $\mathbf{R}^1 = \text{H}$

1b:  $\mathbf{K} = \text{H}$ ,  $\mathbf{R}^1 = \text{H}$

2:  $\mathbf{K} = \alpha\text{-Glc}$ ,  $\mathbf{R}^1 =$

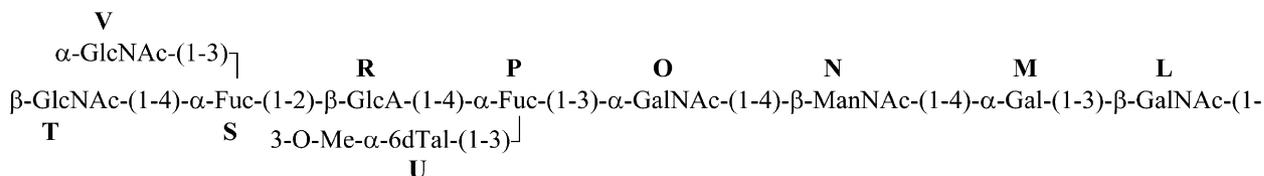


3: - same as 2 with  $\mathbf{R} = \alpha\text{-GlcA}$ .



6  $\mathbf{R}^1 = \text{H}$

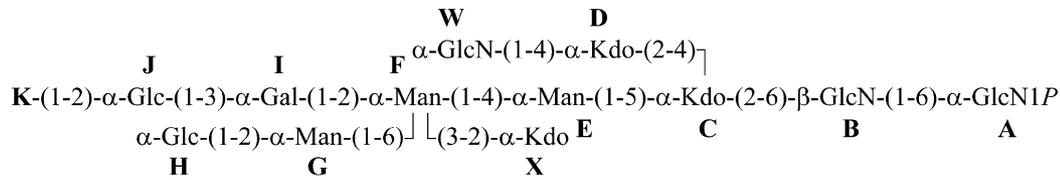
7a  $\mathbf{R}^1 =$



7b: same as 7a but  $\mathbf{U} = 2\text{-O-Ac-}3\text{-O-Me-}\alpha\text{-6dTal}$ ,  $\mathbf{C} = \text{Kdo}$



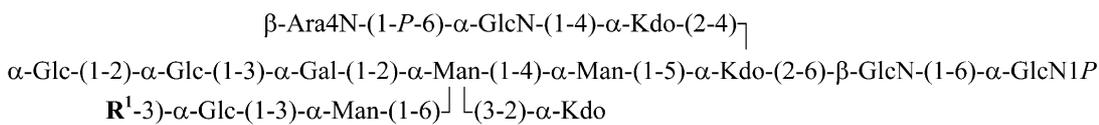
Scheme 1. Structures of the isolated compounds and proposed structure of the carbohydrate backbone of *P. frisingensis* VTT E-82164 LPS.



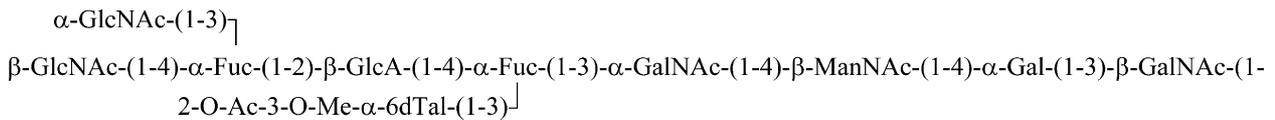
**12a:** K =  $\alpha$ -Glc

**12b:** K = H

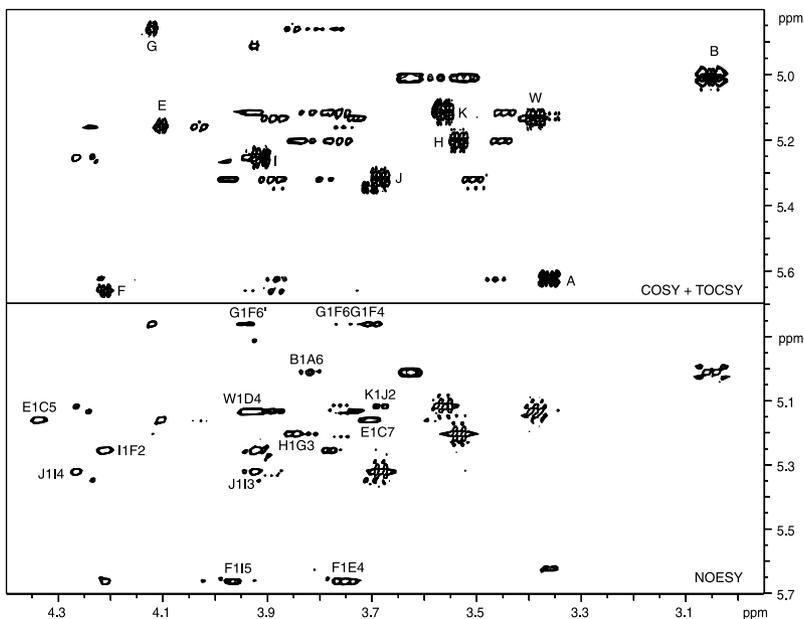
Proposed structure of the carbohydrate backbone of *P. frisingensis* LPS:



where  $\mathbf{R}^1$  is H or



Scheme 1. (Continued).



**Fig. 2.** Sections of COSY, TOCSY, and NOESY spectra of the oligosaccharide **1a**, containing correlations from anomeric protons.

their  $\alpha$ -configuration [16], while the H3 signals of Kdo D appeared at 1.97 (ax) and 2.53 (eq) p.p.m., which may correspond to a  $\beta$ -configuration [16]. However, a NOE correlation observed between H3 of Kdo C and H6 of

Kdo D is possible only in the case of an  $\alpha$ -configuration of residue D, linked to O4 of Kdo C, as follows from molecular modeling. The unusual position of the H3 signals of residue D in product **1a** (as well as in **1b**, **2**, and **3**)

**Table 1.** Assigned NMR spectral data for the isolated oligosaccharides obtained in  $^2\text{H}_2\text{O}$  at 25 °C. Residue nomenclature and oligosaccharide structures are given in Scheme 1.

Unit, compound	Nucleus	1	2 (3eq)	3 (3ax)	4	5	6 (5b)	7 (6b)	8a (OMe)	8b
$\alpha$ -GlcNP A, 2, 3	$^1\text{H}$	5.61	3.35	3.87	3.45	4.13	4.21	3.79		
	$^{13}\text{C}$	91.8	55.6	70.8	70.9	74.1	70.2			
$\beta$ -GlcN B, 2, 3	$^1\text{H}$	4.96	3.04	3.61	3.49	3.61	3.58	3.54		
	$^{13}\text{C}$	100.8	56.8	73.5	71.1	75.7	62.5			
$\alpha$ -Kdo C, 2, 3	$^1\text{H}$		2.04	1.99	4.17	4.32	3.58	3.69	3.89	3.58
	$^{13}\text{C}$		101.5	35.4	70.5	72.7	73.8	70.5	64.8	
Kdo-ol C, 7a,b	$^1\text{H}$		1.98	2.04		3.95	3.90	3.66	3.69	3.87
	$^{13}\text{C}$			38.8		81.6	71.3	72.1	64.2	
$\alpha$ -Kdo D, 2, 3	$^1\text{H}$		2.63	1.91	3.95	4.23	3.74	3.97	3.95	3.81
	$^{13}\text{C}$		100.4	34.9	78.6	66.3	73.0	71.2	64.0	
$\alpha$ -Kdo-ol D, 8	$^1\text{H}$		4.14	2.14/2.08	4.19	4.14	3.75	3.75	3.69	3.86
	$^{13}\text{C}$		71.4	38.0	80.5	71.4	72.1 <sup>a</sup>	73.6 <sup>a</sup>	65.1	
$\alpha$ -GlcN W, 2, 3	$^1\text{H}$	5.12	3.38	3.88	3.34	3.73	3.86	3.78		
	$^{13}\text{C}$	98.5	55.1	70.8	71.1	74.9	62.3			
$\alpha$ -GlcN6P W, 8	$^1\text{H}$	5.43	3.37	3.93	3.62	4.04	4.15	4.21		
	$^{13}\text{C}$	97.4	56.6	71.8	71.3	73.8	66.4			
$\alpha$ -Kdo X, 2, 3	$^1\text{H}$		2.09	1.85	4.07	4.02	3.76	3.93	3.85	3.69
	$^{13}\text{C}$		103.8	35.5	66.6	67.4	73.5	70.3	63.3	
$\alpha$ -Man E, 2, 3	$^1\text{H}$	5.13	4.09	4.01	3.68	4.26	3.74	4.02		
	$^{13}\text{C}$	100.2	71.7	72.3	76.4	71.5	63.6			
$\alpha$ -Man E, 7a,b	$^1\text{H}$	5.07	4.04	4.06	3.83	3.97				
	$^{13}\text{C}$	103.3	72.0	72.0	76.3	73.1				
$\alpha$ -Man F, 2, 3	$^1\text{H}$	5.58	4.20	3.86	3.84	3.75	3.86	3.75		
	$^{13}\text{C}$	101.0	80.3	76.0	66.6	74.0	67.6			
$\alpha$ -Man F, 7a,b	$^1\text{H}$	5.63	4.12	3.96	3.81	3.91	3.79	4.02		
	$^{13}\text{C}$	101.1	82.0	71.7	68.0	72.9	67.1			
$\alpha$ -Man G, 2, 3	$^1\text{H}$	4.84	4.15	3.89	3.81	3.82	3.85	3.76		
	$^{13}\text{C}$	100.6	70.7	81.9	66.9	73.7	62.3			
$\alpha$ -Man G, 7a,b	$^1\text{H}$	4.91	4.16	3.89	3.88	3.78				
	$^{13}\text{C}$	100.8	70.8	81.7	66.9	73.9				
$\alpha$ -Glc H, 2, 3	$^1\text{H}$	5.20	3.78	4.01	3.56	3.87	3.82	3.78		
	$^{13}\text{C}$	102.2	72.8	83.0	68.9	73.4	61.1			
$\alpha$ -Glc H, 7a,b	$^1\text{H}$	5.22	3.62	3.97	3.57	3.90	3.83			
	$^{13}\text{C}$	102.7	72.8	83.2	69.3	72.9	61.4			
$\alpha$ -Gal I, 2, 3	$^1\text{H}$	5.25	3.90	3.91	4.24	3.98	3.73	3.69		
	$^{13}\text{C}$	102.3	68.1	75.3	66.4	73.4	62.3			
$\alpha$ -Gal I, 7a,b	$^1\text{H}$	5.19	4.01	3.95	4.32	4.07				
	$^{13}\text{C}$	102.7	68.3	75.2	66.2	72.6	62.3			
$\alpha$ -Glc J, 2, 3	$^1\text{H}$	5.30	3.68	3.88	3.48	3.95	3.86	3.77		
	$^{13}\text{C}$	93.1	76.7	72.3	70.6	72.1	61.5			
$\alpha$ -Glc J, 7a,b	$^1\text{H}$	5.36	3.69	3.93	3.52	3.98	3.80	3.89		
	$^{13}\text{C}$	92.8	76.6	72.4	70.6	72.7	61.9			
$\alpha$ -Glc K, 2, 3	$^1\text{H}$	5.09	3.55	3.74	3.42	3.93	3.81	3.76		
	$^{13}\text{C}$	97.7	72.3	73.9	70.5	73.0	61.5			
$\alpha$ -Glc K, 7a,b	$^1\text{H}$	5.14	3.60	3.80	3.50	3.96				
	$^{13}\text{C}$	97.6	72.4	74.0	70.5	73.0	61.5			
$\beta$ -GalN L, 2, 3	$^1\text{H}$	4.88	3.38	4.00	4.25	3.71	3.82	3.75		
	$^{13}\text{C}$	101.6	53.5	76.2	64.7	76.3	62.1			
$\beta$ -GalN L, 7a,b	$^1\text{H}$	4.78	4.11	3.86	4.18	3.71				
	$^{13}\text{C}$	103.0	52.3	77.0	65.3	75.9				
$\alpha$ -Gal M, 2, 3	$^1\text{H}$	5.25	3.93	4.09	4.28	3.96	3.73	3.69		
	$^{13}\text{C}$	96.4	69.0	70.1	77.3	72.7	62.3			
$\alpha$ -Gal M, 7a,b	$^1\text{H}$	5.10	3.82	3.90	4.23	3.90				
	$^{13}\text{C}$	97.0	69.1	70.5	77.3	71.8				
$\beta$ -ManN N, 2, 3	$^1\text{H}$	5.11	3.87	4.27	3.79	3.58	3.94	3.83		
	$^{13}\text{C}$	103.0	56.1	71.1	74.3	75.8	61.8			
$\beta$ -ManN N, 7a,b	$^1\text{H}$	4.98	4.61	4.16	3.68	3.53	3.92	3.99		
	$^{13}\text{C}$	100.9	54.8	73.7	76.7	76.1	61.6			

**Table 1.** (Continued).

Unit, compound	Nucleus	1	2 (3eq)	3 (3ax)	4	5	6 (5b)	7 (6b)	8a (OMe)	8b
$\alpha$ -GalN O, <b>2</b>	$^1\text{H}$	5.59	3.66	4.04	4.08	4.07	3.79	3.73		
	$^{13}\text{C}$	99.0	51.2	77.4	69.4	72.6	62.2			
$\alpha$ -GalN O, <b>3</b>	$^1\text{H}$	5.61	3.69	4.06	4.13	4.08	3.79	3.73		
	$^{13}\text{C}$	98.7	51.1	77.7	69.3	72.6	62.2			
$\alpha$ -GalN O, <b>7a,b</b>	$^1\text{H}$	5.22	4.44	3.98	4.04	4.09	3.79	3.83		
	$^{13}\text{C}$	100.9	50.2	75.2	69.9	73.1	62.3			
$\alpha$ -Fuc P, <b>2</b>	$^1\text{H}$	5.11	3.81	4.08	4.04	4.21	1.34			
	$^{13}\text{C}$	98.6	69.5	76.0	80.2	69.1	16.9			
$\alpha$ -Fuc P, <b>3</b>	$^1\text{H}$	5.09	3.89	4.07	4.08	4.26	1.30			
	$^{13}\text{C}$	103.0	73.2	76.0	80.0	69.4	17.1			
$\alpha$ -Fuc P, <b>7a,b</b>	$^1\text{H}$	5.10	3.71	4.08	4.02	4.17	1.39			
	$^{13}\text{C}$	101.9	69.4	76.1	80.6	68.4	16.9			
$\beta$ -GlcA R, <b>2</b>	$^1\text{H}$	4.62	3.68	3.71	3.71	3.61				
	$^{13}\text{C}$	102.9	76.3	78.3	72.0	79.4				
$\beta$ -GlcA R, <b>7a,b</b>	$^1\text{H}$	4.72	3.68	3.78	3.70	3.86				
	$^{13}\text{C}$	103.2	75.4	78.1	72.5	77.0	173.3			
$\alpha$ -GlcA R, <b>3</b>	$^1\text{H}$	5.02	3.75	3.86	3.82	4.39				
	$^{13}\text{C}$	100.5	74.2	70.8	71.6	72.2				
$\alpha$ -6dTal U, <b>2</b>	$^1\text{H}$	5.07	4.05	3.57	3.93	4.23	1.18		3.47	
	$^{13}\text{C}$	104.2	68.0	75.0	70.3	68.6	16.6		56.1	
$\alpha$ -6dTal U, <b>3</b>	$^1\text{H}$	5.10	4.06	3.51	3.90	4.33	1.221		3.45	
	$^{13}\text{C}$	104.0	68.2	75.3	70.3	68.8	16.7		56.2	
$\alpha$ -6dTal U, <b>7b</b>	$^1\text{H}$	5.08	5.24	3.67	3.82	4.21	1.22		3.44	
	$^{13}\text{C}$	102.8	68.5	74.8	69.2	68.2	16.7		56.8	
$\alpha$ -6dTal U, <b>7a</b>	$^1\text{H}$	5.13	4.14	3.55	3.82	4.20	1.24		3.51	
	$^{13}\text{C}$	104.0	67.6	75.3	70.9	68.4	16.6		56.1	
<b>9</b>	$^1\text{H}$	4.83	4.01	3.54	3.95	4.00	1.29		3.46	
<b>9</b> , $J_{n,n+1}$ , Hz		1	3.5	3.5	~1	6.6				
<b>9</b>	$^{13}\text{C}$	102.5	68.1	75.7	70.4	68.2	16.6		56.1	
<b>10</b>	$^1\text{H}$	4.48	4.13	3.46	3.88	3.69	1.32		3.56	
	$^{13}\text{C}$	102.7	68.9	78.6	70.0	72.7	16.5		57.9	
$\alpha$ -11	$^1\text{H}$	5.24	4.00	3.61	3.95	4.18	1.27		3.45	
	$^{13}\text{C}$	95.6	68.8	75.2	70.4	68.1	16.7		56.3	
$\beta$ -11	$^1\text{H}$	4.78	4.06	3.48	3.88	3.70	1.30		3.45	
	$^{13}\text{C}$	95.0	69.6	78.5	69.5	72.3	16.6		56.3	
$\alpha$ -Fuc S, <b>2</b>	$^1\text{H}$	5.56	4.05	4.08	4.20	4.56	1.27			
	$^{13}\text{C}$	99.4	69.2	73.6	78.8	68.0	17.4			
$\alpha$ -Fuc S, <b>3</b>	$^1\text{H}$	5.38	4.03	4.11	4.24	4.39	1.25			
	$^{13}\text{C}$	98.9	69.2	74.1	78.8	68.5	17.2			
$\alpha$ -Fuc S, <b>7a,b</b>	$^1\text{H}$	5.50	4.01	4.09	4.08	4.56	1.29			
	$^{13}\text{C}$	99.6	69.4	72.7	79.0	67.9	17.4			
$\beta$ -GlcN T, <b>2,3</b>	$^1\text{H}$	4.77	3.02	3.58	3.43	3.43	3.93	3.76		
	$^{13}\text{C}$	100.7	57.8	74.3	70.8	77.8	61.5			
$\beta$ -GlcN T, <b>7a,b</b>	$^1\text{H}$	4.71	3.85	3.62	3.44	3.47	3.76	4.04		
	$^{13}\text{C}$	102.2	57.2	74.9	71.6	77.3	62.1			
$\alpha$ -GlcN V, <b>2</b>	$^1\text{H}$	5.43	3.34	3.88	3.57	3.88	3.82	3.75		
	$^{13}\text{C}$	97.8	55.3	73.2	70.3	73.2	61.1			
$\alpha$ -GlcN V, <b>3</b>	$^1\text{H}$	5.44	3.33	3.91	3.51	3.88	3.82	3.75		
	$^{13}\text{C}$	97.8	55.3	73.4	70.6	73.2	61.1			
$\alpha$ -GlcN V, <b>7a,b</b>	$^1\text{H}$	5.24	3.96	3.84	3.66	3.90				
	$^{13}\text{C}$	99.3	54.7	72.5	70.4	72.8				
$\beta$ -Ara4N Y, <b>8</b>	$^1\text{H}$	5.56	3.79	4.21	3.76	3.88	4.26			
	$^{13}\text{C}$	97.6	70.1	67.5	54.1	61.5				

<sup>a</sup> Assignments might be interchanged.

was probably due to its substitution by an  $\alpha$ -GlcN residue. A similar effect was observed for the products obtained from *Acinetobacter* LPS [17,18]. Indeed, the configuration of Kdo D was unambiguously determined

on the basis of NMR analysis of oligosaccharide **4**, in which Kdo D was not substituted and its H3 signals appeared at 1.70 and 1.94 p.p.m., corresponding to an  $\alpha$ -configuration.

The position of the Kdo residue X was identified on the basis of the NOE correlation between its H6 and H2 of the Man residue F (which is analogous to the NOE between protons C3 and D6). This conclusion was confirmed by the results of the methylation analysis of compound **4**. The methylated oligosaccharide was hydrolyzed, and the monosaccharides converted into alditol acetates with deuterium label at C1 using NaBD<sub>4</sub> reduction, acetylated, and analyzed by GC-MS, which allowed identification of all partially methylated alditol acetates expected for structure **4**.

The <sup>31</sup>P-NMR spectrum of **1a** contained only one signal at 2 p.p.m., correlating with H1 of the α-GlcN residue A, with a coupling constant of 6.5 Hz. Thus oligosaccharide **1a** was phosphorylated at A1.

The negative ion mode ES mass spectrum of **1a** gave a molecular mass of 2378 Da, which corresponded to the expected composition Hex<sub>7</sub>HexN<sub>3</sub>Kdo<sub>3</sub>P<sub>1</sub>.

The minor product **1b** contained one hexose residue less than **1a** according to the mass spectrum (molecular mass of 2216 Da, Hex<sub>6</sub>HexN<sub>3</sub>Kdo<sub>3</sub>P<sub>1</sub>). This is confirmed in the NMR data by the absence of the glucose residue K, consistent with the structures shown in Scheme 1.

Oligosaccharides **2** and **3** were isolated in a mixture at a ratio of about 5 : 1. Analysis of the major series of signals in the NMR spectra of this mixture led to the identification of all components of oligosaccharide **1a** and also 10 monosaccharide spin systems (Fig. 3). The NMR spectra of this product were complex, but, at 800 MHz with the use of the standard 2D techniques DQFCOSY, TOCSY, NOESY, HSQC, HMBC, HSQC-TOCSY, HSQC-NOESY, the signal spread was sufficient for identification of all monosaccharides and linkages between them, as presented in Scheme 1. The most problematic assignment was related to the group of signals near 5.1 p.p.m., belonging to Man N, Fuc P, 3-O-methyl-6-deoxytalose U (from **3**), GlcN W, and Glc K. Assignment of the signals of residue N and determination of its position in the structure was possible using <sup>1</sup>H-<sup>13</sup>C correlation spectra (HSQC, HMBC, HSQC-

TOCSY, HSQC-NOESY). The monosaccharide sequence was deduced from the observed transglycosidic correlations from proton-NOE to proton(s)/HMBC to carbon: B1-A6/A6; E1-C5,C7,D7/C5; W1-D4,D5/D4; I1-F2,X6/F2; F1-E4/E4; F2-X6; G1-F6/F6; H1-G2,G3/G3; J1-I3,I4/I3; K1-J2,I4/J2; L1-H3/H3; M1-L3,L4/L3; N1-M4/M4; P1-O3/O3; R1-P4/P4; U1-P3/-; S1-R2/R2; T1-S4/S4; V1-S3/S3. Determination of the substitution position of glucuronic acid R was difficult because of extensive overlapping of its <sup>1</sup>H and <sup>13</sup>C NMR signals. It was found to be substituted at O2 from the methylation analysis and from the data for other oligosaccharides. The problems with residues N, R, U were resolved in the analysis of the oligosaccharide **7**, which showed no signal overlap for the corresponding residues. In general, all assignments were confirmed by methylation analysis.

The residue 3-O-methyl-6-deoxyhexose (U) had all small intra-ring coupling constants (< 3 Hz) in the <sup>1</sup>H-NMR spectrum, which could correspond to an α-*talo*- or an α-*gulo*- configuration. For the reliable determination of its configuration, a model compound methyl 3-O-methyl-6-deoxy-α-D-talopyranoside (**9**), and its β-anomer (**10**), were synthesized. This was achieved by configuration inversion at C2 and C4 in the methyl 3-O-methyl-2,4-di-O-trifluoromethylsulfonyl-6-deoxy-α,β-D-glucopyranoside. NMR data (<sup>1</sup>H and <sup>13</sup>C chemical shifts and vicinal coupling constants) for the synthetic compound **9** were close to those of the residue U in the oligosaccharides (Table 1). Monosaccharides were furthermore identified by GC as alditol acetates. Thus the residue of 3-O-methyl-6-deoxyhexose had a *talo*-configuration. 3-O-Methyl-6-deoxy-D-talopyranose, **11**, was isolated from the hydrolysate of the LPS. It contained α-pyranose and β-pyranose anomeric forms (NMR data in Table 1), and a smaller amount of furanoside forms (data for furanoses not presented).

In addition, a minor series of signals in the spectra of the **2** + **3** mixture could be attributed to structure **3**, with a single difference from **2** to an altered anomeric

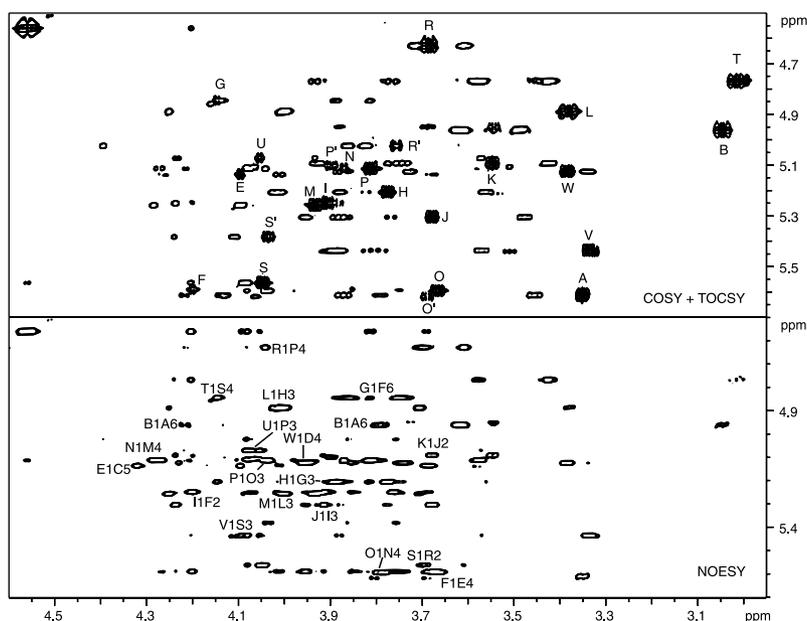


Fig. 3. Sections of COSY, TOCSY, and NOESY spectra of the mixture of the oligosaccharides **2** (letter labels) and **3** (letters with apostrophe labels), containing correlations from anomeric protons.

configuration of the residue of GlcA R, being  $\alpha$  in **3**. The origin of  $\alpha$ -GlcA is not clear; it was not present among the products of mild acid hydrolysis and thus may be an artefact of alkaline treatment.

Structures **2** and **3** were in agreement with ESI-MS data, which determined a molecular mass of 3973.5 Da (Hex<sub>8</sub>-HexN<sub>8</sub>HexA<sub>1</sub>dHex<sub>3</sub>Kdo<sub>3</sub>P<sub>1</sub>Me<sub>1</sub>).

Methylation analysis of the O-deacylated LPS was performed using the Ciucanu-Kerek method [12]. Methylated product was converted into a mixture of partially methylated alditol acetates by acid hydrolysis, reduction with NaBD<sub>4</sub>, and acetylation. On another sample, the methylated product was depolymerized by acid methanolysis, treated with NaBD<sub>4</sub> to reduce carboxy groups, hydrolysed, reduced with NaBD<sub>4</sub>, and acetylated. The second procedure led to the reduction of the GlcA residue with the introduction of two deuterium labels at C6. Comparison of the two chromatograms allowed unambiguous confirmation that GlcA is substituted at position 2. The substitution positions of all the other monosaccharides were confirmed by GC-MS data of the methylated products to be as presented in Scheme 1.

Deamination of the products of complete deacylation of the LPS led to the oligosaccharides **4** and **5**, representing undecasaccharide and pentasaccharide fragments of oligosaccharides **1a** and/or **2**. These products were isolated by HPAEC (after borohydride reduction) and analysed by NMR spectroscopy, ESI MS, and methylation. The most important result obtained from NMR analysis of compound **4** was the determination of the anomeric configuration of Kdo D (see above).

Mild acid hydrolysis of the LPS with subsequent borohydride reduction and separation of the products by HPAEC in alkaline buffer led to the isolation of three main compounds **6**, **7a**, and **8**. The 18-residue oligosaccharide **7a** contained all the components of oligosaccharide **2**, except the Kdo residues D and X, GlcN residues A, B, and W. All amino sugars were N-acetylated. NMR spectra of this oligosac-

charide were analysed (Table 1) and found to be consistent with the structure presented in Scheme 1. Especially useful for the assignment was the well-separated position of the H1 signal of ManN N, which allowed unambiguous determination of its anomeric configuration as  $\beta$ , based on the intraresidual NOE between H1 and H3,5 (all axial) and the low-field position of its C5 at 76.1 p.p.m. No  $\alpha$ -GlcA was found in the products, thus we conclude that  $\alpha$ -GlcA in product **3** was a result of configuration inversion during strong alkaline treatment. ESI MS data confirmed the structure of **7a** (observed mass of 3181 Da) and showed that it contained minor amount of the structure with missing hexose. As in products **1–3**, Glc residue K was missing.

Analysis of the oligosaccharide **7b**, obtained after mild acid hydrolysis by gel chromatography without reduction, showed that it contains an O-acetyl group on O2 of 3-O-Me- $\alpha$ -6dTal residue U. Acetylation of O2 led to low-field shift of the residue U H2 signal to 5.24 p.p.m. (compare with 4.14 p.p.m. in **7a**). Its C1 signal was shifted 2 p.p.m. to high field in **7b** compared with **7a** (Table 1) because of the  $\beta$ -effect of the acetylation. Acetylation of **7b** was confirmed by ESI MS data, which gave the expected mass of 3221.4 Da.

The spectra of oligosaccharide **6** were completely assigned, and its structure was determined as presented in Scheme 1 (NMR data not shown). A variant of **6** without Glc K was also isolated as a minor compound.

The product **8** contained the residue of 4-amino-4-deoxyarabinose (Y), which was not found in the products of alkaline deacylation of the LPS. It was linked to position 6 of the GlcN residue by a phosphodiester bond (<sup>31</sup>P signal at -0.3 p.p.m., correlating with H6 of GlcN and H1 of Ara4N). The residue of Ara4N1P was lost after KOH deacylation and therefore was not present in oligosaccharides **1–3**.

The NOE spectra of oligosaccharides **2**, **3**, and **7a,b** contained a number of correlations from H6 of 6-deoxysugars (Fig. 4). This fact was used as additional proof of the structural assignment. The terminal heptasaccharide fragment including monosaccharide residues from O to V was

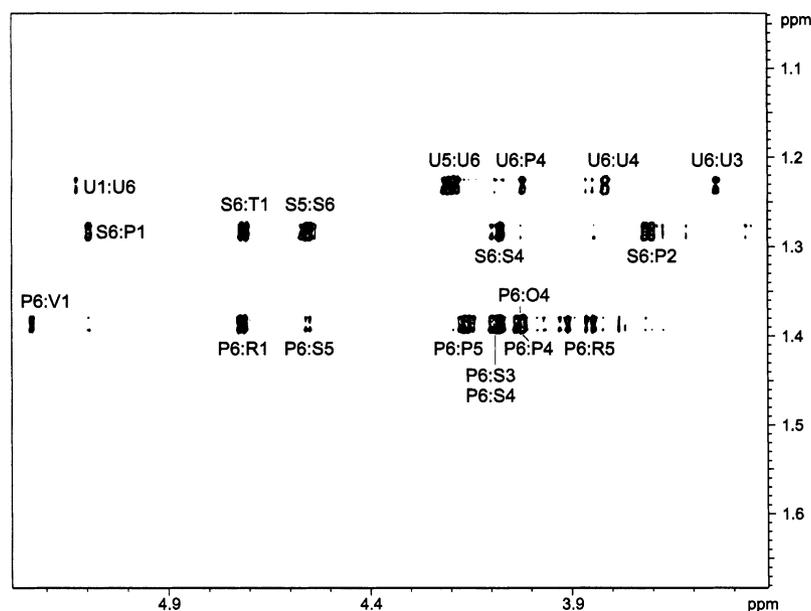


Fig. 4. Part of the NOESY spectrum of compound **7b**, containing correlations of H6 of 6-deoxysugars.



## Acknowledgements

This work was supported by the Canadian Bacterial Diseases Network. We thank Donald Krajcarsky (NRC Canada) for the ESI MS analysis. The spectra at 800 MHz were obtained on the Varian Unity Inova spectrometer of the Danish Instrument Center for NMR Spectroscopy of Biological Macromolecules.

## References

- Lee, S.Y., Mabee, M.S. & Jangaard, N.O. (2002) *Pectinatus*, a new genus of the family Bacteroidaceae. *Int. J. Syst. Bacteriol.* **28**, 582–594.
- Haikara, A. & Helander, I. (1992) *Pectinatus*, *Megasphaera* and *Zymophilus*. In *The Prokaryotes. A Handbook on the Biology of Bacteria: Ecophysiology, Isolation, Identification, Applications* (Balows, A., Truper, H.G., Dworkin, M.S., Harder, W. & Schleifer, K.-H., eds), Springer Verlag.
- Suihko, M.-L. & Haikara, A. (2001) Characterization of *Pectinatus* and *Megasphaera* strains by automated ribotyping. *J. Inst. Brew.* **107**, 175–184.
- Helander, I.M., Hurme, R., Haikara, A. & Moran, A.P. (1992) Separation and characterization of two chemically distinct lipopolysaccharides in two *Pectinatus* species. *J. Bacteriol.* **174**, 3348–3354.
- Senchenkova, S.N., Shashkov, A.S., Moran, A.P., Helander, I.M. & Knirel, Y.A. (1995) Structures of the O-specific polysaccharide chains of *Pectinatus cerevisiiphilus* and *Pectinatus frisingensis* lipopolysaccharides. *Eur. J. Biochem.* **232**, 552–557.
- Helander, I.M., Kilpeläinen, I., Vaara, M., Moran, A.P., Lindner, B. & Seydel, U. (1994) Chemical structure of the lipid A component of lipopolysaccharides of the genus *Pectinatus*. *Eur. J. Biochem.* **224**, 63–70.
- Helander, I.M., Moll, H. & Zähringer, U. (1993) 4-O-(2-amino-2-deoxy- $\alpha$ -D-glucopyranosyl)-3-deoxy-D-manno-2-octulosonic acid, a constituent of lipopolysaccharides of the genus *Pectinatus*. *Eur. J. Biochem.* **213**, 377–381.
- Brade, H. & Galanos, C. (1982) Isolation, purification, and chemical analysis of the lipopolysaccharide and lipid A of *Acinetobacter calcoaceticus* NCTC 10305. *Eur. J. Biochem.* **122**, 233.
- Kjaer, M., Andersen, K.V. & Poulsen, F.M. (1994) Automated and semiautomated analysis of homo- and heteronuclear multidimensional nuclear magnetic resonance spectra of proteins: the program PRONTO. *Methods Enzymol.* **239**, 288–308.
- Tsai, C.M. & Frasch, C.E. (1982) A sensitive silver stain for detecting lipopolysaccharides in polyacrylamide gels. *Anal. Biochem.* **119**, 115–119.
- Vinogradov, E.V., Cedzynski, M., Ziolkowski, A. & Swierzko, A. (2001) The structure of the core region of the lipopolysaccharide from *Klebsiella pneumoniae* O3. 3-Deoxy- $\alpha$ -D-manno-octulosonic acid ( $\alpha$ -Kdo) residue in the outer part of the core, a common structural element of *Klebsiella pneumoniae* O1, O2, O3, O4, O5, O8, and O12 lipopolysaccharides. *Eur. J. Biochem.* **268**, 1722–1729.
- Ciucanu, I. & Kerek, F. (1984) A simple and rapid method for the permethylation of carbohydrates. *Carbohydr. Res.* **131**, 209–217.
- Holst, O., Thomas-Oates, J.E. & Brade, H. (1994) Preparation and structural analysis of oligosaccharide monophosphates obtained from lipopolysaccharide of recombinant strains of *Salmonella mimesota* and *Escherichia coli*, expressing the genus-specific epitope of *Chlamydia* lipopolysaccharide. *Eur. J. Biochem.* **222**, 183–194.
- Kapur, B.M. & Allgeier, H. (1968) 3-O-Methyl-6-deoxy-L-talose, Synthese und Identifizierung mit L-Acovenose. *Helv. Chim. Acta* **51**, 89–94.
- Duus, J.Ø., Gotfredsen, C.H. & Bock, K. (2000) Carbohydrate structural determination by NMR spectroscopy: modern methods and limitations. *Chem. Rev.* **100**, 4589–4614.
- Birnbaum, G.I., Roy, R., Brisson, J.R. & Jennings, H.J. (1987) Conformations of ammonium 3-deoxy-D-manno-2-octulosonate (KDO) and methyl  $\alpha$ - and  $\beta$ -ketopyranosides of KDO: X-ray structure and  $^1\text{H}$  NMR Analyses. *J. Carbohydr. Chem.* **6**, 17–39.
- Vinogradov, E.V., Duus, J.Ø., Brade, H. & Holst, O. (2002) The structure of the carbohydrate backbone of the lipopolysaccharide from *Acinetobacter baumannii* strain ATCC 19606. *Eur. J. Biochem.* **269**, 422–430.
- Vinogradov, E.V., Petersen, B., Thomas-Oates, J., Brade, H. & Holst, O. (1998) Characterization of a novel branched tetrasaccharide of 3-deoxy-D-manno-oct-2-ulopyranosonic acid. The structure of the carbohydrate backbone of the lipopolysaccharide from *Acinetobacter baumannii* strain NCTC 10303 (ATCC 17904). *J. Biol. Chem.* **273**, 28122–28131.
- Lipkind, G.M., Shashkov, A.S., Knirel, Y.A., Vinogradov, E.V. & Kochetkov, N.K. (1988) A computer-assisted structural analysis of regular polysaccharides on the basis of  $^{13}\text{C}$ -NMR data. *Carbohydr. Res.* **175**, 59–75.
- Holst, O. (1999) Chemical structure of core region of lipopolysaccharides. In *Endotoxin in Health and Disease* (Brade, H., Morrison, D.C., Opal, S.M. & Vogel, S., eds), pp. 115–154. Marcel Dekker Inc., New York.
- Forsberg, L.S., Bhat, U.R. & Carlson, R.W. (2000) Structural characterization of the O-antigenic polysaccharide of the lipopolysaccharide from *Rhizobium etli* Strain CE3. A unique O-acetylated glycan of discrete size, containing 3-O-methyl-6-deoxy-L-talose and 2,3,4-tri-O-methyl-L-fucose. *J. Biol. Chem.* **275**, 18851–18863.
- Weckesser, J., Drews, G. & Mayer, H. (1979) Lipopolysaccharides of photosynthetic prokaryotes. *Annu. Rev. Microbiol.* **33**, 215–239.